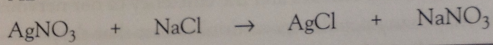
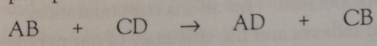


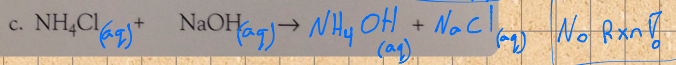
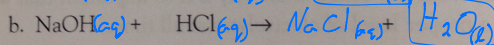
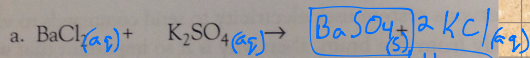
Warm-up:

Double Up

A double-replacement reaction is a reaction in which two ionic compounds exchange ions to create two new ionic compounds. Such reactions often take place when the reactants are in the form of an aqueous solution. In some cases, one or more of the resulting products is insoluble in water and settles out as a solid called a precipitate. For example:



Complete the following double-replacement reactions.



Net Ionic Equations: a chemical equation that only shows ions undergoing change

